

Synthesis, characterization, and antimicrobial evaluation of some novel Imidazo[1,2-c] pyrimidine derivatives

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ABSTRACT: Due to chalcone compounds' diverse and intriguing bioactive characteristics, the present work recommends the production of several new chalcone derivatives, followed by spectral characterization utilizing various spectral techniques such as IR, NMR, and elemental analyses. In addition to, assessment the antimicrobial activity of imidazo[1,2-c] pyrimidine derivatives. All tested new chalcone compounds showed antibacterial activity against *P. aeruginosa* and *B. subtilis*, while only compounds (5a-d and 5f) displayed activity against *E. coli*. Furthermore, only compounds (5a-c, 5e and 5f) reduced the development of *S. aureus*. The compound 5f revealed significant antibacterial activity against *S. aureus* (87.5%), *B. subtilis* (82.61 %), *P. aeruginosa* (78.26%) and *E. coli* (64%), when compared to Ampicillin. On the other hand, the compounds (5a-d, 5e and 5f) showed antifungal activity against *C. Albicans*. Also, compounds (5a, 5b, 5c, 5d and 5f) displayed activity against *A. flavus*. A significant antifungal activity for compound 5b has been observed against *C. Albicans* (70.37%) and *A. flavus* (52%), when compared to Colitrima-zole.

KEYWORDS: new chalcones, imidazo[1,2-c] pyrimidine, antibacterial activity, antifungal activity

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I. INTRODUCTION

Microbial infections, particularly bacterial infections, have become more widespread in recent decades, causing a high risk of mortality among immunocompromised people. Opportunistic bacterial infections represent a considerable risk to these people and are occurring at an alarming pace [1, 2]. Since there have been several reports on the isolation of bacteria that are known to be sensitive to commonly used antibiotics and have become multiresistant to other treatments on the market [3, 4].

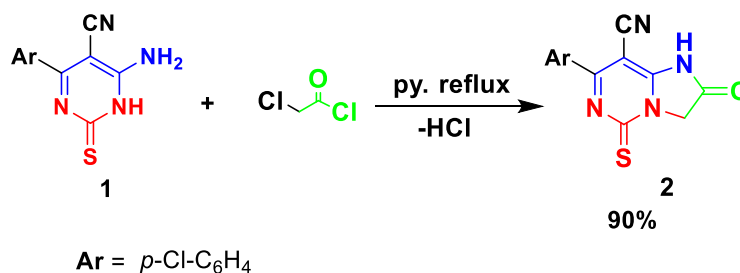
Undoubtedly, heterocyclic molecules, especially those with nitrogen atom, have important biological functions [5, 6]. Imidazopyrimidine scaffolds are important building blocks in drug design and development. Imidazopyrimidines, for example, have a diverse pharmacological profile, including anticancer [7], antitubercular [8], antiviral [9], antimicrobial [10], antifungal [11], anti-inflammatory [12], parasiticidal activity [13], calcium channel blockers [14], benzodiazepine receptor agonists [15], potent P38 MAP kinase inhibitors [16] and GABA receptor ligands [17]. It has also been used as an azodye [18], fabric whitener, insecticidal, acaricidal, and nematocidal agent [19]. It was also discovered to be a significant structural component of Divaplon [20] and Fasiplon [15] as a possible anxiolytic and anticonvulsant medication whose use in clinical practise was discontinued [21]. To expand its uses, this scaffold can also be employed as an organic fluorophore in biomarkers and photochemical sensors [22].

In agreement with our previous interest in the synthesis of imidazopyrimidine derivatives [23-25], and because of the resulting pharmacological interest in compounds that belong to these heterocyclic derivatives, we would like to report a method for preparing some novel Imidazo[1,2-c] pyrimidine derivatives, as well as, an evaluation of their antimicrobial activity.

II. RESULTS AND DISCUSSION

Chemistry:

The current investigation began with the synthesis and characterization of 2-thioxo cytosine **1** which was then cyclized with chloroacetyl chloride to yield the imidazo[1,2-*c*] pyrimidine **2** as described in the literature [24] (Scheme 1). Compound **2** has α -CH acid function at C-3 and operates as an active site for the formation of new bioactive chemicals *via* Kneovenagel condensation.



Scheme 1: Synthesis of Imidazo[1,2-*c*] pyrimidine **2**.

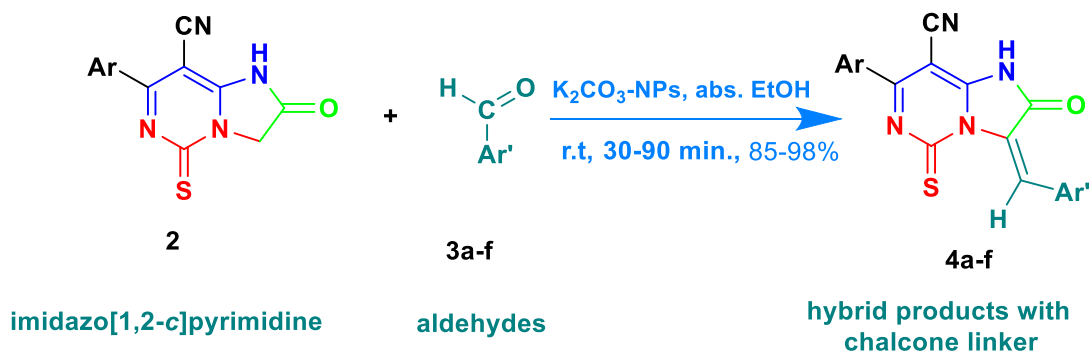
Condensation with an aromatic or hetero aromatic aldehyde in ethanol at room temperature for 30-90 minutes in the presence of K₂CO₃-NPs functionalizes C-H (sp³) at the third position in imidazo[1,2-*c*]pyrimidine **3** to benzylidene function. (Scheme 2). A variety of aromatic aldehydes were utilised (Table 1), including 4-hydroxybenzaldehyde, 2-methoxybenzaldehyde, 2-tolualdehyde, 3-nitrobenzaldehyde, 4-cyanobenzaldehyde, and 3-hydroxy-4-methoxybenzaldehyde. High yield (up to 96%) and short reaction time with aldehyde-containing electron withdrawing groups (e.g., CN, NO₂) (30 min.) were found. The aldehyde-containing electron-donating group (e.g., OH, OCH₃, CH₃), on the other hand, yields a high yield (87-98%).

The IR spectra of the synthesized compounds **4a-f** include four main bands in the ranges 3185-3371, 2190-2194, 1656-1697, and 1211-1317 cm⁻¹ for the N-H, C≡N, C=O, and C=S functions, respectively.

The presence of a benzyldene proton as a singlet in the range $\delta_{\text{H}} = 4.61$ -6.79 ppm and the absence of a CH₂-imidazole ring in the ¹H-NMR spectrum of compounds **4a-f** are the main evidence promoting the condensation reaction and the synthesis of the target molecules. In addition, the aromatic component of aldehyde appeared in the expected field, and their ¹³C-NMR corresponded to the chemical structure.

The ¹H-NMR chart of imidazopyrimidine **4a** assigned three singlets at δ_{H} 6.79, 10.49 and 11.51 ppm for benzyldene (=CH), OH and NH protons, respectively. Furthermore, at δ_{H} 7.22, 7.44, 7.63 and 7.71 ppm, the two aryl rings split into four doublets. Its ¹³C-NMR chart recorded 16 signals for sp³, sp², and sp carbons at δ_{C} 52.45, 113.9, 114.9, 118.3, 118.5, 126.0, 126.8, 127.3, 134.3, 135.8, 142.8, 152.4, 158.2, 163.5, 164.8 and 171.5 ppm.

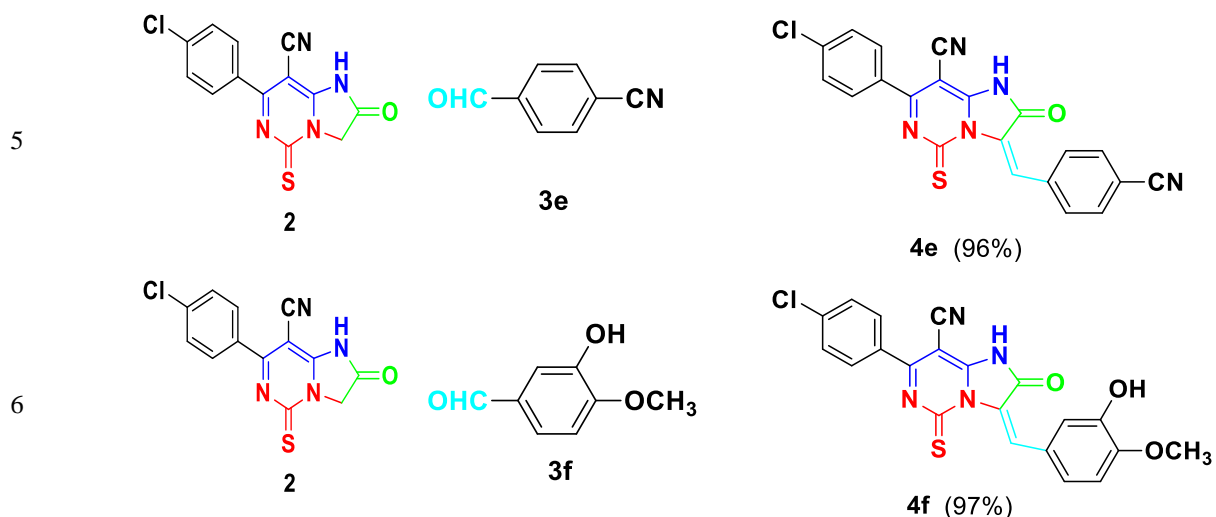
The existence of O-H, N-H, C≡N, C=O, and C=S functions in the IR data of **4f** validates its structure at 3353, 3185, 2190, 1697, and 1317 cm⁻¹, respectively. The imidazopyrimidine **4f** ¹H-NMR spectrum revealed four singlets at δ_{H} 3.94, 6.31, 12.60 and 12.99 for methoxy group, benzyldene (=CH), OH and N-H protons, respectively. Furthermore, at δ_{H} 7.13, 7.23-7.33 and 7.38 ppm, the two aryl rings divide into two doublets and one multiplet. The **4f** ¹³C-NMR spectra accounted for 19 signals at 52.07, 57.01, 118.9, 119.7, 121.9, 122.2, 123.5, 124.4, 125.6, 129.1, 129.5, 129.7, 130.0, 131.5, 146.4, 151.3, 162.6, 166.4 and 173.0 ppm for sp³, sp² and sp carbons.



Scheme 2: Formation of chalcone linker via condensation reaction between imidazo[1,2-c]pyrimidine and aldehydes.

Table 1: Chemical structure of imidazo[1,2-c]pyrimidine and aldehydes fragments and the hybrid products.

Entry	imidazopyrimidine	Aldehydes	Hybrid products (yield %)
1			 4a (98%)
2			 4b (87%)
3			 4c (92%)
4			 4d (85%)



Assessment antimicrobial activity:

The antimicrobial activity of the synthesized compounds **4a-f** was tested against two gram-positive bacteria (*Staphylococcus aureus*, *Bacillus subtilis*), two gram-negative bacteria (*Escherichia coli*, *Pseudomonas aeruginosa*), and two fungi (*Candida albicans*, *Aspergillus flavus*) obtained from the Faculty of Pharmacy, Mansoura University, Egypt. The disc diffusion technique, as reported in the literature [26], was employed to assess the antimicrobial activity of synthesized compounds. In this study, the results were compared to the activity of commercially available conventional antibiotics (Ampicillin and Colitrimazole). The results are shown in (Table 2).

Table (2): Antimicrobial Activities of compounds (4a-f).

Compound	<i>E. coli</i>		<i>P.aeruginosa</i>		<i>S. aureus</i>		<i>B. subtilis</i>		<i>C. Albicans</i>		<i>A. flavus</i>	
	Diameter of inhibition zone (mm)	% Activity index	Diameter of inhibition zone (mm)	% Activity index	Diameter of inhibition zone (mm)	% Activity index	Diameter of inhibition zone (mm)	% Activity index	Diameter of inhibition zone (mm)	% Activity index	Diameter of inhibition zone (mm)	% Activity index
4a	9	36	12	52.17	13	54.16	11	47.82	12	44.44	8	32
4b	12	48	15	65.22	17	70.83	16	69.57	19	70.37	13	52
4c	8	32	13	56.52	15	62.5	15	65.22	16	59.26	10	40
4d	6	24	4	17.39	NA	----	5	21.74	10	37.04	7	28
4e	NA	----	6	26.09	3	12.5	4	17.39	5	18.52	NA	----
4f	16	64	18	78.26	21	87.5	19	82.61	17	62.96	12	48
Ampicillin	25	100	23	100	24	100	23	100	NA	----	NA	----
Colitrimazole	NA	----	NA	----	NA	----	NA	----	27	100	25	100

- NA → No Activity.

According to the results were recorded in Table (2), the synthesized compounds showed antibacterial activity against *P. aeruginosa* and *B. subtilis*, while only compounds (**4a-d** and **4f**) displayed activity against *E. coli*. Furthermore, only compounds (**4a-c**, **4e** and **4f**) reduced the development of *S. aureus*. With regards to *E. coli*, **4f** displayed the highest antibacterial activity (16 mm), followed by **4b** (12 mm), **4a** (9 mm), **4c** (8 mm) and **4d** (6 mm). Also, **4f** revealed significant antibacterial activity against *P. aeruginosa* (18 mm) compared to **4b** (15 mm), **4c** (13 mm), **4a** (12 mm), **4e** (6 mm) and **4d** (4 mm). Only compounds (**4f**, **4b**, **4c**, **4a** and **4e**) displayed activity against *S. aureus* (21, 17, 15, 13 and 3 mm), respectively. On the other hand, the compounds (**4f**, **4b**, **4c**, **4a**, **4d** and **4e**) exhibited antibacterial activity against *B. subtilis* (19, 16, 15, 11, 5 and 4 mm), respectively.

Furthermore, the compounds (**4b**, **4f**, **4c**, **4a**, **4d** and **4e**) had antifungal activity against *C. Albicans* (19, 17, 16, 12, 11, 10 and 5 mm), respectively. Also, compounds (**4b**, **4f**, **4c**, **4a** and **4d**) displayed activity against *A. flavus* (13, 12, 10, 8 and 7 mm), respectively.

Moreover, **4b** had strong antibacterial activity against *P. aeruginosa* (65.22%), *S. aureus* (70.83%) and *B. subtilis* (69.57 %), and moderate against *E. coli* (48%), while **4c** showed strong antimicrobial activity against *S. aureus* (62.5%) and *B. subtilis* (65.22%) and moderate against *E. coli* (32%), *P. aeruginosa* (56.52%). Furthermore, **4a** showed moderate antibacterial activity against *E. coli* (36%), *P. aeruginosa* (52.17%), *S. aureus* (54.16%) and *B. subtilis* (47.82%). Also, **4d** showed weak antibacterial activity against *E. coli* (24%), *P. aeruginosa* (17.39%) and *B. subtilis* (21.74%). With regards to **4e** showed weak antimicrobial activity against *P. aeruginosa* (26.09 %), *S. aureus* (12.5 %) and *B. subtilis* (17.39%). In addition, **4f** had strong antibacterial activity against *E. coli* (64%), *P. aeruginosa* (78.26%), *S. aureus* (87.5%) and *B. subtilis* (82.61 %) when compared to Ampicillin.

With regards to *C. Albicans*, the compounds (**4b**, **4c** and **4f**) showed strong antifungal activity (70.37 %, 59.26 % and 62.96%), respectively, while the compounds (**4a**, **4d** and **4e**) showed weak antifungal activity (44.44 %, 37.04% and 18.52%), respectively. Also, the compounds (**4b** and **4f**) showed moderate antifungal activity against *A. flavus* (52 % and 48%), respectively, while the compounds (**4c**, **4a** and **4d**) showed weak antifungal activity (40%, 32 % and 28%), respectively, when compared to Colitriza-zole.

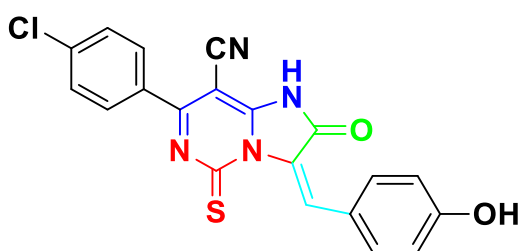
III. Experimental

Melting points were determined using an Electro-thermal IA 9100 equipment. On a Nexus 670 FTIR Nicolet, Fourier transform infrared spectrometer, IR spectra (KBr) were obtained. The ^1H and ^{13}C NMR spectra were measured in DMSO- d_6 using a JEOL-JNM-LA 400 MHz spectrometer for ^1H NMR and a JEOL-JNM-LA 100 MHz spectrometer for ^{13}C NMR. TMS is the industry standard for expressing chemical changes on the (ppm) scale. TLC was detected using UV light on Merck Silica Gel 60F254. At Cairo University's Microanalysis Centre, element analysis was done using a PerkinElmer 240 (microanalysis). The chemicals and solvents used in the synthesis of imidazo[1,2-*c*]pyrimidine derivatives were given by Merck and Aldrich. Compounds **2** and **3** are synthesized and analyzed in the same manner as stated before [24, 25].

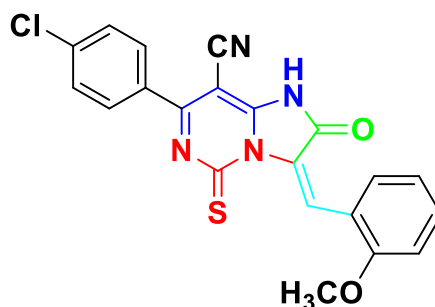
General method for synthesis of target imidazo[1,2-*c*]pyrimidines 4a-f

A solution of imidazole **2** (0.01 mol) and aromatic aldehyde **3** namely, 4-hydroxybenzaldehyde, 2-methoxybenzaldehyde, 2-tolualdehyde, 3-nitrobenzaldehyde, 4-cyanobenzaldehyde and 3-hydroxy-4-methoxybenzaldehyde (0.01 mol) in absolute ethanol was stirred at room temperature for 30-90 minutes (TLC indicator) in the presence of K_2CO_3 -NPs (0.01 mol). After the reaction was completed, the mixture was cooled and diluted with cold H_2O , and the resulting product was filtered, dried, and recrystallized from ethanol.

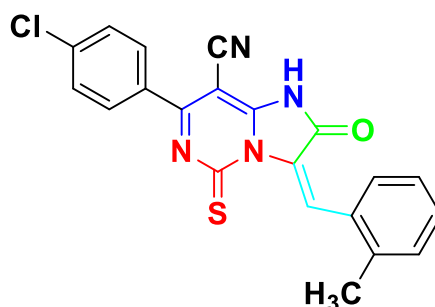
7-(4-Chlorophenyl)-3-(4-hydroxybenzylidene)-2-oxo-5-thioxo-1,2,3,5-tetrahydroimidazo[1,2-*c*]pyrimidine-8-carbonitrile (**4a**)



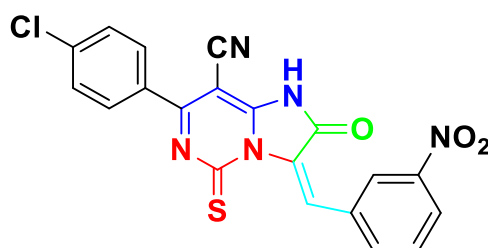
Yellow powder, mp 308-310 °C; yield: 98 %. IR (ν_{max} , cm^{-1}): 3450 (O-H), 3245 (N-H), 2193 ($\text{C}\equiv\text{N}$), 1676 ($\text{C}=\text{O}$), 1298 ($\text{C}=\text{S}$). ^1H NMR (400 MHz, DMSO- d_6): δ_{H} = 6.79 (s, 1H, =CH), 7.22 (d, 2H, $^3J = 8.40$ Hz, H_{aryl}), 7.44 (d, 2H, $^3J = 8.80$ Hz, H_{aryl}), 7.63 (d, 2H, $^3J = 8.80$ Hz, H_{aryl}), 7.71 (d, 2H, $^3J = 7.20$ Hz, H_{aryl}), 10.49 (s, 1H, OH, exchange with D_2O), 11.51 (s, 1H, NH, exchange with D_2O). ^{13}C NMR, δ = 52.45, 113.9, 114.9, 118.3, 118.5, 126.0, 126.8, 127.3, 134.3, 135.8, 142.8, 152.4, 158.2, 163.5, 164.8 and 171.5. Anal. Calcd for $\text{C}_{20}\text{H}_{11}\text{ClN}_4\text{O}_2\text{S}$ (406.84): C, 59.04; H, 2.73; N, 13.77 %. Found: C, 58.95; H, 2.75; N, 13.82

7-(4-Chlorophenyl)-3-(2-methoxybenzylidene)-2-oxo-5-thioxo-1,2,3,5-tetrahydroimidazo[1,2-c]pyrimidine-8-carbonitrile (4b)

Yellow powder, mp 305-307 °C; yield: 87 %. IR (ν_{\max} , cm^{-1}): 3320 (N-H), 2191 (C≡N), 1656 (C=O), 1301 (C=S). ^1H NMR (400 MHz, DMSO- d_6): δ_{H} = 3.92 (s, 3H, OCH₃), 6.58 (s, 1H, =CH), 7.38 (t, 2H, 3J = 7.20 Hz, H_{aryl}), 7.42 (d, 2H, 3J = 7.60 Hz, H_{aryl}), 7.51 (d, 2H, 3J = 6.80 Hz, H_{aryl}), 7.60 (d, 2H, 3J = 9.60 Hz, H_{aryl}), 10.99 (s, 1H, NH, exchange with D₂O). ^{13}C NMR, δ = 52.62, 54.82, 114.7, 115.8, 120.0, 126.6, 128.9, 130.2, 130.8, 131.8, 141.7, 148.2, 154.2, 159.8, 164.3 and 171.7. Anal. Calcd for C₂₁H₁₃ClN₄O₂S (420.87): C, 59.93; H, 3.11; N, 13.31 %. Found: C, 59.82; H, 3.09; N, 13.37.

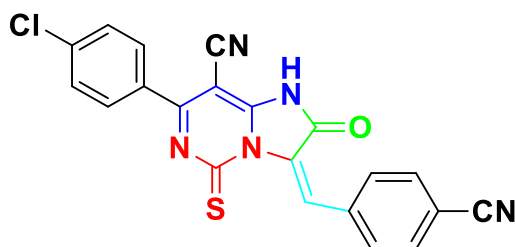
7-(4-Chlorophenyl)-3-(2-methylbenzylidene)-2-oxo-5-thioxo-1,2,3,5-tetrahydroimidazo[1,2-c]pyrimidine-8-carbonitrile (4c)

Pale yellow powder, mp 302-304 °C; yield: 92 %. IR (ν_{\max} , cm^{-1}): 3185 (N-H), 2191 (C≡N), 1658 (C=O), 1211 (C=S). ^1H NMR (400 MHz, DMSO- d_6): δ_{H} = 2.69 (s, 3H, CH₃), 4.61 (s, 1H, =CH), 7.17 (t, 1H, 3J = 8.40 Hz, H_{aryl}), 7.29 (t, 1H, 3J = 8.00 Hz, H_{aryl}), 7.44 (d, 2H, 3J = 8.00 Hz, H_{aryl}), 7.64 (d, 2H, 3J = 8.40 Hz, H_{aryl}), 7.76 (d, 2H, 3J = 8.00 Hz, H_{aryl}), 10.97 (s, 1H, NH, exchange with D₂O). ^{13}C NMR, δ = 25.12, 50.92, 115.5, 120.4, 128.9, 129.1, 130.1, 130.5, 131.6, 135.1, 140.0, 144.3, 151.1, 160.0, 163.1 and 171.1. Anal. Calcd for C₂₁H₁₃ClN₄OS (404.87): C, 62.30; H, 3.24; N, 13.84 %. Found: C, 62.21; H, 3.27; N, 13.89.

7-(4-Chlorophenyl)-3-(3-nitrobenzylidene)-2-oxo-5-thioxo-1,2,3,5-tetrahydroimidazo[1,2-c]pyrimidine-8-carbonitrile (4d)

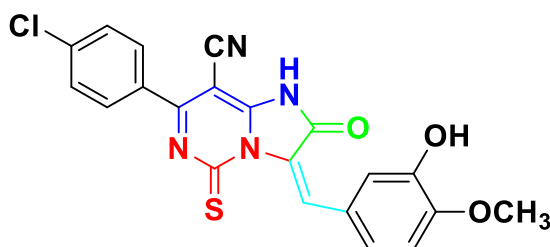
Pale brown powder, mp 311-313 °C; yield: 85 %. IR (ν_{\max} , cm^{-1}): 3236 (N-H), 2192 (C≡N), 1692 (C=O), 1292 (C=S). ^1H NMR (400 MHz, DMSO- d_6): δ_{H} = 5.17 (s, 1H, =CH), 7.03 (s, 1H, H_{aryl}), 7.11 (t, 1H, 3J = 8.40 Hz, H_{aryl}), 7.21 (d, 2H, 3J = 8.80 Hz, H_{aryl}), 7.43 (d, 2H, 3J = 9.60 Hz, H_{aryl}), 7.62 (d, 2H, 3J = 7.60 Hz, H_{aryl}), 9.57 (s, 1H, NH, exchange with D₂O). ^{13}C NMR, δ = 52.42, 115.4, 120.4, 123.3, 131.3, 131.4, 132.5, 132.8, 133.9, 135.0, 144.4, 148.4, 149.9, 150.3, 160.8, 163.3 and 175.5. Anal. Calcd for C₂₀H₁₀ClN₅O₃S (435.84): C, 55.12; H, 2.31; N, 16.07 %. Found: C, 55.21; H, 2.28; N, 16.14.

7-(4-Chlorophenyl)-3-(4-cyanobenzylidene)-2-oxo-5-thioxo-1,2,3,5-tetrahydroimidazo[1,2-c]pyrimidine-8-carbonitrile (4e)



Yellow powder, mp 314-316 °C; yield: 96 %. IR (ν_{\max} , cm^{-1}): 3371 (N-H), 2194 ($\text{C}\equiv\text{N}$), 1666 ($\text{C}=\text{O}$), 1296 ($\text{C}=\text{S}$). ^1H NMR (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 5.09$ (s, 1H, =CH), 7.38 (d, 2H, $^3J = 7.20$ Hz, H_{aryl}), 7.42 (d, 2H, $^3J = 7.60$ Hz, H_{aryl}), 7.44 (d, 1H, $^3J = 7.60$ Hz, H_{aryl}), 7.46 (d, 1H, $^3J = 7.20$ Hz, H_{aryl}), 7.51 (d, 1H, $^3J = 7.20$ Hz, H_{aryl}), 7.61 (d, 1H, $^3J = 8.40$ Hz, H_{aryl}), 11.41 (s, 1H, NH, exchange with D_2O). ^{13}C NMR, $\delta = 55.95$, 114.8, 117.1, 128.3, 129.4, 130.4, 130.6, 131.8, 142.0, 142.1, 142.2, 151.4, 157.3, 161.4, 168.6 and 171.6. Anal. Calcd for $\text{C}_{21}\text{H}_{10}\text{ClN}_5\text{OS}$ (415.86): C, 60.65; H, 2.42; N, 16.84 %. Found: C, 60.72; H, 2.45; N, 16.91.

7-(4-Chlorophenyl)-3-(3-hydroxy-4-methoxybenzylidene)-2-oxo-5-thioxo-1,2,3,5-tetrahydroimidazo[1,2-c]pyrimidine-8-carbonitrile (4f)



Yellow powder, mp 318-320 °C; yield: 97 %. IR (ν_{\max} , cm^{-1}): 3353 (O-H), 3185 (N-H), 2190 ($\text{C}\equiv\text{N}$), 1697 ($\text{C}=\text{O}$), 1317 ($\text{C}=\text{S}$). ^1H NMR (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 3.94$ (s, 3H, OCH_3), 6.31 (s, 1H, =CH), 7.13 (d, 2H, $^3J = 7.60$ Hz, H_{aryl}), 7.23-7.33 (m, 3H, H_{aryl}), 7.38 (d, 2H, $^3J = 8.40$ Hz, H_{aryl}), 12.60 (s, 1H, OH, exchange with D_2O), 12.99 (s, 1H, NH, exchange with D_2O). ^{13}C NMR, $\delta = 52.07$, 57.01, 118.9, 119.7, 121.9, 122.2, 123.5, 124.4, 125.6, 129.1, 129.5, 129.7, 130.0, 131.5, 146.4, 151.3, 162.6, 166.4 and 173.0. Anal. Calcd for $\text{C}_{21}\text{H}_{13}\text{ClN}_4\text{O}_3\text{S}$ (436.87): C, 57.74; H, 3.00; N, 12.82 %. Found: C, 57.82; H, 3.10; N, 12.80.

IV. REFERENCES

- [1] Wall, G., & Lopez-Ribot, J. L. (2020). Current antimycotics, new prospects, and future approaches to antifungal therapy. *Antibiotics*, 9(8), 445.
- [2] Akinyemi, K. O., Oluwa, O. K., & Omomigbehin, E. O. (2006). Antimicrobial activity of crude extracts of three medicinal plants used in south-west Nigerian folk medicine on some food borne bacterial pathogens. *African Journal of Traditional, Complementary and Alternative Medicines*, 3(4), 13-22.
- [3] Sakagami Y. and Kajimura K. (2002). Bactericidal activities of disinfectants against vancomycin-resistant enterococci. *J. Hosp. Infec.*, 50 (2): 140-144.
- [4] Mahesh B. and Satish S. (2008). Antimicrobial activity of some important medicinal plant against plant and human pathogens. *World Journal of Agricultural Sciences*, 4 (S): 839-843.
- [5] AbdEl-Azim, M. H., Aziz, M. A., Mouneir, S. M., EL-Faragy, A. F., and Shehab, W. S. (2020): Ecofriendly synthesis of pyrano [2, 3-d] pyrimidine derivatives and related heterocycles with anti-inflammatory activities. *Archiv der Pharmazie*, 353(9), 2000084.
- [6] Shehab, W. S., EL-Faragy, A. F., Abdelhamid, A. O., and Aziz, M. A. (2019): Synthesis and biological application of pyranopyrimidine derivatives catalyzed by efficient nanoparticles and their nucleoside analogues. *Synthetic Communications*, 49(24), 3560-3572.
- [7] Aeluri, R., Alla, M., Polepalli, S., & Jain, N. (2015). Synthesis and antiproliferative activity of imidazo [1, 2-a] pyrimidine Mannich bases. *European Journal of Medicinal Chemistry*, 100, 18-23.

- [8] Margiotta, N., Ostuni, R., Ranaldo, R., Denora, N., Laquintana, V., Trapani, G., & Natile, G. (2007). Synthesis and characterization of a platinum (II) complex tethered to a ligand of the peripheral benzodiazepine receptor. *Journal of medicinal chemistry*, 50(5), 1019-1027.
- [9] Gueiffier, A., Lhassani, M., Elhakmaoui, A., Snoeck, R., Andrei, G., Chavignon, O., & Chapat, J. P. (1996). Synthesis of acyclo-C-nucleosides in the imidazo [1, 2-*a*] pyridine and pyrimidine series as antiviral agents. *Journal of medicinal chemistry*, 39(14), 2856-2859.
- [10] Rival, Y., GRASSY, G., & MICHEL, G. (1992). Synthesis and antibacterial activity of some Imidazo [1, 2-*a*] pyrimidine derivatives. *Chemical and pharmaceutical bulletin*, 40(5), 1170-1176.
- [11] Rival, Y., Grassy, G., Taudou, A., & Ecalle, R. (1991). Antifungal activity in vitro of some imidazo [1, 2-*a*] pyrimidine derivatives. *European journal of medicinal chemistry*, 26(1), 13-18.
- [12] Vidal, A., Ferrándiz, M. L., Ubeda, A., Acero-Alarcón, A., Sepulveda-Arques, J., & Alcaraz, M. J. (2001). Effect of imidazo [1, 2-*a*] pyrimidine derivatives on leukocyte function. *Inflammation Research*, 50(6), 317-320.
- [13] Aeluri, R., Alla, M., Polepalli, S., & Jain, N. (2015). Synthesis and antiproliferative activity of imidazo [1, 2-*a*] pyrimidine Mannich bases. *European Journal of Medicinal Chemistry*, 100, 18-23.
- [14] Sanfilippo, P. J., Urbanski, M., Press, J. B., Dubinsky, B., & Moore Jr, J. B. (1988). Synthesis of (aryloxy) alkylamines. 2. Novel imidazo-fused heterocycles with calcium channel blocking and local anesthetic activity. *Journal of medicinal chemistry*, 31(11), 2221-2227.
- [15] Tully, W. R., Gardner, C. R., Gillespie, R. J., & Westwood, R. (1991). 2-(Oxadiazolyl)-and 2-(thiazolyl) imidazo [1, 2-*a*] pyrimidines as agonists and inverse agonists at benzodiazepine receptors. *Journal of medicinal chemistry*, 34(7), 2060-2067.
- [16] Fekri, L. Z., Nikpassand, M., & Khakshoor, S. N. (2019). Green, effective and chromatography free synthesis of benzoimidazo [1, 2-*a*] pyrimidine and tetrahydrobenzo [4, 5] imidazo [1, 2-*d*] quinazolin-1 (2H)-one and their pyrazolyl moiety using Fe₃O₄@ SiO₂@ l-proline reusable catalyst in aqueous media. *Journal of Organometallic Chemistry*, 894, 18-27.
- [17] Humphries, A. C., Gancia, E., Gilligan, M. T., Goodacre, S., Hallett, D., Merchant, K. J., & Thomas, S. R. (2006). 8-Fluoroimidazo [1, 2-*a*] pyridine: Synthesis, physicochemical properties and evaluation as a bioisosteric replacement for imidazo [1, 2-*a*] pyrimidine in an allosteric modulator ligand of the GABAA receptor. *Bioorganic & medicinal chemistry letters*, 16(6), 1518-1522.
- [18] Xie, Y. Y. (2005). Organic Reactions in Ionic Liquids: Ionic Liquid-Accelerated One-Pot Synthesis of 2-Arylimidazo [1, 2-*a*] pyrimidines. *Synthetic communications*, 35(13), 1741-1746.
- [19] Goel, R., Luxami, V., & Paul, K. (2015). Synthetic approaches and functionalizations of imidazo [1, 2-*a*] pyrimidines: an overview of the decade. *RSC advances*, 5(99), 81608-81637.
- [20] Feely, M., Boyland, P., Picardo, A., Cox, A., & Gent, J. P. (1989). Lack of anticonvulsant tolerance with RU 32698 and Ro 17-1812. *European journal of pharmacology*, 164(2), 377-380.
- [21] Atack, J. R. (2005). The benzodiazepine binding site of GABAA receptors as a target for the development of novel anxiolytics. *Expert opinion on investigational drugs*, 14(5), 601-618.
- [22] Velázquez-Olvera, S., Salgado-Zamora, H., Velázquez-Ponce, M., Campos-Aldrete, E., Reyes-Arellano, A., & Pérez-González, C. (2012). Fluorescent property of 3-hydroxymethyl imidazo [1, 2-*a*] pyridine and pyrimidine derivatives. *Chemistry Central Journal*, 6(1), 1-9.
- [23] Prasad, P., Kalola, A. G., & Patel, M. P. (2018). Microwave assisted one-pot synthetic route to imidazo [1, 2-*a*] pyrimidine derivatives of imidazo/triazole clubbed pyrazole and their pharmacological screening. *New Journal of Chemistry*, 42(15), 12666-12676.
- [24] Mohammed, S. M., Moustafa, A. H., El-Sayed, H. A., Amin, A. S., Haggag, A., Tantawy, E. S., & Haggam, R. A. (2022). A New Approach for Synthesis of 2-Thioxocytosine-5-carbonitrile as Antimicrobial Agents. *Russian Journal of General Chemistry*, 92(9), 1806-1813.
- [25] Mohammed, S. M., Moustafa, A. H., El-Sayed, H. A., & Gad, E. M. (2023). Cytotoxic Activity and Docking Study of Some 3-Arylidine Imidazo [1, 2-*c*] Pyrimidine Candidates Synthesized Via CH (sp³) Functionalization. *Synthetic Communications*.
- [26] Stylianakis, I., Kolocouris, A., Kolocouris, N., Fytas, G., Foscolos, G. B., Padalko, E., & De Clercq, E. (2003). Spiro [pyrrolidine-2, 2'-adamantanes]: synthesis, anti-influenza virus activity and conformational properties. *Bioorganic & medicinal chemistry letters*, 13(10), 1699-1703.